

IRON'S IMPACT ON NITROGEN CYCLING IN AN OPEN WATER WETLAND TO

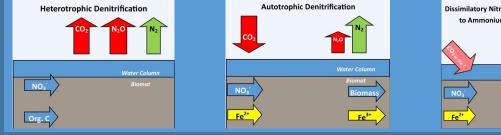
PROMOTE SUSTAINABLE WATER TREATMENT

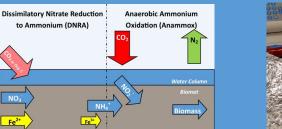


Re-Inventing the Nation's Urban Water Infrastructure (ReNUWIt)

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Background Methods Microorganisms in constructed wetlands anaerobically remove nitrate Anaerobic slurry microcosms with wetland biomat, differing concentrations from the water. of ferrous iron given at 15 days Nitrate removal can take the form of various pathways including Sampling: ٠ denitrification, DNRA, anammox, and abiotic pathways. Aqueous: nitrogen species, sulfide, iron, organic and inorganic • Microorganisms can use organic carbon (heterotrophy) or inorganic carbon, acetate carbon (autotrophy). Gas: nitrous oxide and methane The role of ferrous iron in these processes was previously unknown. Mineral: ferric oxides Microbial: functional genes for nitrogen cycling processes **Hypothesis** Heterotrophic denitrification will dominate. Secondarily, low iron concentrations will experience autotrophic denitrification. High iron concentrations will cause a shift towards DNRA (and anammox).

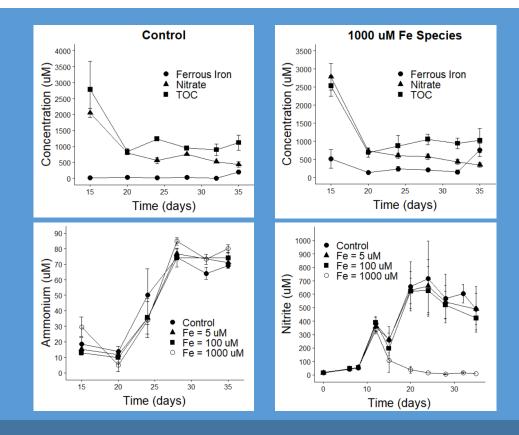






Results

- Reduction in nitrate that corresponds with reduction in organic carbon
- No significant difference in ammonium production among iron concentrations
- More nitrite reduction with high concentration of ferrous iron
- Little difference between control, 5 ppm, and 100 ppm iron microcosms with all species



Conclusions

• Hypothesis:

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•Heterotrophic denitrification will dominate. Supported.

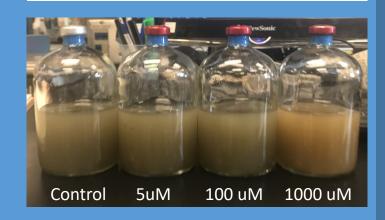
•Secondarily, low concentrations will experience autotrophic denitrification.

No Evidence.

• High iron concentrations will cause a shift towards DNRA (and anammox).

Refuted.

Likely that abiotic nitrite reduction occurs beyond a threshold iron concentration.



Unfoiled microcosms at the final time point. The red hue in the high iron condition indicates iron oxidation